DEPARTMENT OF CHEMISTRY

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casey.case.edu
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John D. Protasiewicz, Department Chair
john.protasiewicz@case.edu

The Department of Chemistry is the largest department representing the chemical sciences at Case Western Reserve University. It consists of 19 faculty members, 15 associated faculty, about 5 postdoctoral associates, approximately 60 graduate students, and over 100 undergraduate students majoring in chemistry. The department offers undergraduate and graduate degree programs leading to the Bachelor of Arts, Bachelor of Science, Master of Science, and Doctor of Philosophy.

The general focus of chemistry is on (1) understanding the basic properties of matter, and (2) employing this knowledge in the design, synthesis, and characterization of materials with novel and useful properties. The various degree programs strive to develop all aspects of the student’s chemical knowledge through a broad range of lecture and laboratory courses.

Chemical research is an integral part of the department’s activities: over $3 million of federal, state, and private research support flows into the department each year. State-of-the-art research facilities are available to both graduate and undergraduate students. Undergraduates are encouraged to participate in research projects with individual faculty members in order to expand their hands-on training, problem-solving skills, and understanding of the scientific method as applied in chemical research. These research projects typically involve interchange and collaboration across all levels of experience and may also involve scientists from other departments and institutions.

Chemistry is often referred to as “the central science” because of its key role in interdisciplinary studies. Correspondingly, a degree in chemistry affords a broad range of employment opportunities. Chemists can direct their talents to specialized problems of applied research, or they can choose to delve into fundamental investigations. A degree in chemistry can cover the spectrum of chemical specialties, from biochemistry to interstellar chemistry. The degree also provides valuable preparation for other professions, such as medicine, dentistry, and law.

The American Chemical Society (http://www.acs.org), with more than 160,000 members, is the major professional society in the United States for practicing chemists. Both undergraduate and graduate students may join the society.

Facilities

The department’s facilities for experimental and theoretical research are modern and extensive. They include diverse major instruments for use by faculty and students, as well as specialized equipment serving individual research groups. Shared instrumentation includes 400- and 600-MHz NMR spectrometers, ultrafast laser systems in the Center for Chemical Dynamics, and a cyber-enabled X-ray crystallographic facility.

Other departmental instrumentation includes equipment for laser Raman spectroscopy, GC-MS and LC-MS/MS mass spectrometers, calorimeters, stopped-flow kinetics instrumentation, a circular dichroism spectrometer, an analytical ultracentrifuge, and equipment for electrochemical measurements. Access to very high-field NMR instrumentation is available on campus at the Cleveland Center for Membrane Structural Biology (CCMSB), which is equipped with numerous 500- to 900-MHz NMR spectrometers for solution and solid-state measurements. The chemistry department’s computers are part of the campus-wide fiber optic communications network operated by Information Technology Services, and the entire University Circle area offers wireless access. In addition to the full complement of software, Internet, and library database services offered by the university, connections to off-site databases, such as SciFinder and Ohio Supercomputer Center, are available to departmental users.

The department uses some of the foremost equipment available in high-resolution nuclear magnetic resonance spectroscopy and in tunable laser spectroscopy. Work on various aspects of chemistry as studied by these techniques is recognized throughout the world.

Primary Faculty

John D. Protasiewicz, PhD
(Cornell University)
Hurlbut Professor of Chemistry and Chair
Inorganic chemistry, materials and energy, organometallic chemistry, photochemistry, catalysis, computational chemistry, crystallography, electrochemistry, green chemistry, main group chemistry, molecular electronics, nanotechnology, OLEDs, optoelectronics, physical organic chemistry, polymers, solar energy, solid-state chemistry, spectroscopy, supramolecular chemistry, synthesis

Clemens Burda, PhD
(University of Basel, Switzerland)
Chemical Professor
Photochemistry, materials, physical chemistry, nanochemistry, bio- and energy applications, biophysical and biomedical science and engineering, spectroscopy

Carlos E. Crespo-Hernández, PhD
(University of Puerto Rico)
Professor
Analytical chemistry, biophysical chemistry, energy, photochemistry, physical chemistry, chemical dynamics and kinetics, computational chemistry, environmental chemistry, time-resolved spectroscopy

Thomas G. Gray, PhD
(Harvard University)
Professor
Organometallic chemistry, inorganic chemistry, energy

Irene Lee, PhD
(Pennsylvania State University)
Professor
Biochemistry, medicinal chemistry, bioorganic chemistry

Fu-Sen Liang, PhD
(The Scripps Research Institute)
Assistant Professor
Bioorganic chemistry

Drew A. Meyer, PhD
(Stanford University)
John Teagle Professorial Fellow in Chemistry, Senior Instructor
Physical chemistry, inorganic chemistry, X-ray spectroscopy, chemical education

Other research faculty focuses include:

John Teagle Professorial Fellow in Chemistry
Bioorganic chemistry

Thomas G. Gray, PhD
(Harvard University)
Professor
Organometallic chemistry, inorganic chemistry, energy

Irene Lee, PhD
(Pennsylvania State University)
Professor
Biochemistry, medicinal chemistry, bioorganic chemistry

Fu-Sen Liang, PhD
(The Scripps Research Institute)
Assistant Professor
Bioorganic chemistry

Drew A. Meyer, PhD
(Stanford University)
John Teagle Professorial Fellow in Chemistry, Senior Instructor
Physical chemistry, inorganic chemistry, X-ray spectroscopy, chemical education
Shane M. Parker, PhD  
(Northwestern University)  
**Assistant Professor**  
Computational and theoretical chemistry

Anthony J. Pearson, PhD  
(University of Aston, Birmingham, England)  
*Rudolph and Susan Rense Professor of Chemistry*  
Organic chemistry, organometallic chemistry, catalysis, natural products, synthesis

Robert G. Salomon, PhD  
(University of Wisconsin, Madison)  
*Charles Frederic Mabery Professor of Research in Chemistry*  
Biochemistry, chemical biology, medicinal chemistry, organic chemistry, bioorganic chemistry, cellular biology, molecular biology, natural products, pharmacology, synthesis

Anna C. Samia, PhD  
(Georgia Institute of Technology)  
**Associate Professor**  
Analytical chemistry, inorganic chemistry, materials and energy, bioinorganic chemistry, functional nanomaterials, nanotheranostics

Geneviève Sauvé, PhD  
(California Institute of Technology)  
**Associate Professor**  
Materials and energy, organic chemistry, physical chemistry, functional polymers, nanoscale morphology, organic electronics, solar energy conversion, structure-property relationships

Daniel A. Scherson, PhD  
(University of California, Davis)  
*Frank Hovorka Professor of Chemistry*  
Analytical chemistry, materials, physical chemistry, photochemistry, electrochemistry

Rekha R. Srinivasan, PhD  
(Case Western Reserve University)  
*James Stephen Swinehart, PhD, Professorial Teaching Fellow in Chemistry; Senior Instructor*  
Analytical chemistry, biophysical chemistry, organic chemistry, chemical education

Gregory P. Tochtrop, PhD  
(Washington University Medical School)  
**Professor**  
Biochemistry, biophysical chemistry, chemical biology, medicinal chemistry, organic chemistry, bioorganic chemistry, synthesis

Blanton S. Tolbert, PhD  
(University of Rochester)  
**Professor**  
Biochemistry, biophysical chemistry, structural biology

Michael G. Zagorski, PhD  
(Case Western Reserve University)  
**Professor**  
Biochemistry, biophysical chemistry, chemical biology, organic chemistry, beer brewing, bioorganic chemistry, drug delivery, NMR, structural biology

**Lecturer**

Krista Cunningham, PhD  
(The Ohio State University)  
**Full-time Lecturer**  
Organic chemistry, chemical education

**Research Faculty**

Mikhail D. Linetsky, PhD  
(Academy of Science of Ukraine)  
**Research Professor**  
Biochemistry, chemical biology, protein chemistry, post-translational protein modification, proteomics

**Secondary Faculty**

Rigoberto Advincula, PhD  
(University of the Philippines)  
**Professor**  
Macromolecular science and engineering

Paul Carey, PhD  
(University of Sussex, UK)  
**Professor, Department of Biochemistry**  
Biochemistry, biophysical chemistry, microscopy/imaging, spectroscopy

John W. Crabb, PhD  
(University of Kansas Medical Center)  
**Professor, Department of Cell Biology, Lerner Research Institute, Cleveland Clinic**  
Proteomics of the visual cycle and age-related ocular diseases

Chris Dealwis, PhD  
**Associate Professor, Department of Pharmacology**  
Biochemistry, biophysics, enzyme catalysis, pharmacology, proteins

Thomas Gerken, PhD  
(Case Western Reserve University)  
**Professor, Division of Pediatric Pulmonology**  
Biochemistry, biophysical chemistry, chemical biology, glycosylation, protein chemistry, protein structure

Michael Harris, PhD  
(University of Alabama at Birmingham)  
**Associate Professor**  
Biochemistry

Thomas Kelley, PhD  
(University of Notre Dame)  
**Associate Professor, Division of Pediatric Pulmonology**  
Biochemistry, medicinal chemistry, cellular biology, pharmacology

David Schiraldi, PhD  
(University of Oregon)  
**Professor, Department of Macromolecular Science & Engineering**  
Polymer synthesis and structure-property relationships, condensation polymers, polymer nanocomposites, fuel cell durability, polymerization catalysis, transport phenomena and packaging applications, polymer blends and complex polymer systems
Adjunct Faculty

Ormond Brathwaite, PhD  
(City University of New York)  
*Adjunct Associate Professor*  
Biochemistry

James Burgess, PhD  
(Longwood College)  
*Adjunct Professor*  
Bioinorganic chemistry, electrochemistry

Michael J. Kenney, PhD  
(Iowa State University)  
*Adjunct Associate Professor*  
Analytical chemistry, physical chemistry, chemical education, computer programming, application development

M. Cather Simpson, PhD  
(University of New Mexico)  
*Adjunct Associate Professor*  
Biophysical chemistry; spectroscopic studies of biologically significant processes

Emeritus Faculty

Alfred B. Anderson, PhD  
(Johns Hopkins University)  
*Emeritus Professor*  
Materials, physical chemistry, electrocatalysis, interfacial phenomena, catalysis, theoretical chemistry

Mary D. Barkley, PhD  
(University of California, San Diego)  
*Emeritus Professor and M. Roger Clapp University Professor of Arts and Sciences*  
Analytical chemistry, biochemistry, biophysical chemistry, medicinal chemistry, photochemistry, physical chemistry, theoretical chemistry

Malcolm E. Kenney, PhD  
(Cornell University)  
*Emeritus Professor and Hurlbut Professor of Chemistry*  
Biochemistry, inorganic chemistry, materials and energy, organometallic chemistry, bioinorganic chemistry, computational chemistry, drug delivery, dyes and pigments, medicinal chemistry, nanotechnology, organosilicon chemistry, photochemistry, photodynamic therapy, polymers

Bachelor of Science in Chemistry

The Department of Chemistry offers three curricula for undergraduate majors, leading to a Bachelor of Science (BS) degree in chemistry, Bachelor of Arts (BA) degree in chemistry, or Bachelor of Arts (BA) degree in chemical biology.

**Total Units Required for Graduation: 120**

**Chemistry BS - Required Chemistry Courses**

<table>
<thead>
<tr>
<th>Year</th>
<th>Fall</th>
<th>Spring</th>
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<tbody>
<tr>
<td>First Year</td>
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<tr>
<td>Principles of Chemistry I (CHEM 105)</td>
<td>3</td>
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</tr>
<tr>
<td>Principles of Chemistry Laboratory (CHEM 113)</td>
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<td>Principles of Chemistry II (CHEM 106)</td>
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<tr>
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<table>
<thead>
<tr>
<th>Year</th>
<th>Fall</th>
<th>Spring</th>
</tr>
</thead>
<tbody>
<tr>
<td>Second Year</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Quantitative Analysis Laboratory (CHEM 304)</td>
<td>2</td>
<td></td>
</tr>
<tr>
<td>Foundations of Analytical Chemistry (CHEM 310)</td>
<td>3</td>
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<tr>
<td>Organic Chemistry I (CHEM 323)</td>
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<tr>
<td>Laboratory Methods in Organic Chemistry (CHEM 322)</td>
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<td>3</td>
</tr>
<tr>
<td>Organic Chemistry II (CHEM 324)</td>
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### Third Year

<table>
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<th>Units Fall</th>
<th>Units Spring</th>
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<tbody>
<tr>
<td>Inorganic Chemistry I (CHEM 311)</td>
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<tr>
<td>Laboratory Methods in Inorganic Chemistry (CHEM 331)</td>
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</tr>
<tr>
<td>Physical Chemistry I (CHEM 335)</td>
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<tr>
<td>Laboratory Methods in Physical Chemistry (CHEM 332)</td>
<td>3</td>
</tr>
<tr>
<td>Physical Chemistry II (CHEM 336)</td>
<td>3</td>
</tr>
<tr>
<td>Chemistry Elective (300-level, see text below)</td>
<td>3</td>
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<td><strong>Year Total:</strong></td>
<td>9 9</td>
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### Fourth Year

<table>
<thead>
<tr>
<th>Units Fall</th>
<th>Units Spring</th>
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<tbody>
<tr>
<td>Research Requirement:</td>
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<tr>
<td>Undergraduate Research (CHEM 397) or Undergraduate Research/Senior Capstone Project (CHEM 398)</td>
<td>3 - 6</td>
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<tr>
<td>Biochemistry Requirement (one of the following):</td>
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</tr>
<tr>
<td>Introductory Biochemistry I (CHEM 328) (spring, 3 units)</td>
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<tr>
<td>Biochemistry II: Living Systems (CHEM 329) (fall, 3 units)</td>
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<tr>
<td>Introduction to Biochemistry: From Molecules To Medical Science (BIOC 307) (4 units)</td>
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<tr>
<td>Chemistry Elective (300-level, see text below)</td>
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<tr>
<td>Technical Electives (see text below)</td>
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<td><strong>Year Total:</strong></td>
<td>6-10 9</td>
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**Total Units in Sequence:** 55-59

### Chemistry BS - Additional Required Courses

- **MATH 121** Calculus for Science and Engineering I 4
- **MATH 122** Calculus for Science and Engineering II 4
- **MATH 223** Calculus for Science and Engineering III 3
- **MATH 227** Calculus III 3
- One of the following: 3
  - **MATH 224** Elementary Differential Equations
  - **MATH 228** Differential Equations
  - **STAT 312** Basic Statistics for Engineering and Science
- **PHYS 121** General Physics I - Mechanics 4
- **PHYS 122** General Physics II - Electricity and Magnetism 4
- **PHYS 124** Physics and Frontiers I - Mechanics
- **PHYS 221** Introduction to Modern Physics 3

**Total Units** 25

The chemistry elective may be any chemistry department course at the 300 level or above which is not part of the "core set," or selected courses with a strong chemistry content at the 300 level or above from other science departments. Only three units of CHEM 397 may be applied to a chemistry elective.

The technical electives may be chosen more widely from any of the physical sciences, math, or engineering courses. An additional six units of CHEM 397 may be taken as technical electives. Further additional units of CHEM 397 may be taken as free electives. Students may wish to group their electives into "tracks" of specialization in order to tailor their degree to a particular area of chemistry.

BS majors who plan to go on to graduate study may elect to take advanced courses in:

- **Inorganic Chemistry**
  - **CHEM 412** Advanced Inorganic Chemistry I
- **Organic Chemistry**
  - **CHEM 421** Advanced Organic Chemistry I
  - **CHEM 422** Advanced Organic Chemistry II
  - **CHEM 435** Synthetic Methods in Organic Chemistry
- **Physical Chemistry**
  - **CHEM 406** Chemical Kinetics
  - **CHEM 407** Chemical Thermodynamics
  - **CHEM 446** Quantum Mechanics I

Students can also elect to take other graduate offerings. Interdisciplinary strengths can be achieved by selecting technical electives in biochemistry, biomedical engineering, chemical engineering, macromolecular science, and materials science as well as in biology; earth, environmental, and planetary sciences; mathematics, applied mathematics, and statistics; and physics.

### Bachelor of Arts in Chemistry

The BA program in chemistry is intended for pre-professional students who plan careers in medicine, dentistry, veterinary medicine, pharmacy, or in other fields for which a baccalaureate degree in chemistry provides appropriate training. BA majors may supplement their required courses with additional chemistry courses or may utilize the curriculum's flexibility to develop an interdisciplinary program of their choice. Many chemistry BA majors participate in undergraduate research within the Department of Chemistry (CHEM 397 / CHEM 398) or in other science departments, including those in the medical school.

**Total Units Required for Graduation:** 120

### Chemistry BA - Required Chemistry Courses

<table>
<thead>
<tr>
<th>Units</th>
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<tbody>
<tr>
<td>Fall</td>
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<tr>
<td>Principles of Chemistry I (CHEM 105)</td>
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<tr>
<td>Principles of Chemistry Laboratory (CHEM 113)</td>
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<tr>
<td>Principles of Chemistry II (CHEM 106)</td>
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<td><strong>Year Total:</strong></td>
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**Second Year**

<table>
<thead>
<tr>
<th>Units</th>
<th>Fall</th>
<th>Spring</th>
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<tbody>
<tr>
<td>Introductory Organic Chemistry I (CHEM 223) or Organic Chemistry I (CHEM 323)</td>
<td>3</td>
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<tr>
<td>Introductory Organic Chemistry Laboratory I (CHEM 233) (see below*)</td>
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<tr>
<td>Introductory Organic Chemistry II (CHEM 224) or Organic Chemistry II (CHEM 324)</td>
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<td>Introductory Organic Chemistry Laboratory II (CHEM 234) (see below*) or Laboratory Methods in Organic Chemistry (CHEM 322)</td>
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<td>5-6</td>
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* CHEM 322 is offered in spring only, and may be substituted in place of both CHEM 233 and CHEM 234. However, some medical schools require two semesters of organic lab, so students should plan accordingly.

**Third Year**

<table>
<thead>
<tr>
<th>Units</th>
<th>Fall</th>
<th>Spring</th>
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<tbody>
<tr>
<td>Introductory Physical Chemistry I (CHEM 301) or Physical Chemistry I (CHEM 335)</td>
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<td>Quantitative Analysis Laboratory (CHEM 304)</td>
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<td>Foundations of Analytical Chemistry (CHEM 310)</td>
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<td>Introductory Physical Chemistry II (CHEM 302) or Physical Chemistry II (CHEM 336)</td>
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**Fourth Year**

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<th>Units</th>
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<td>Total Units in Sequence:</td>
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* CHEM 322 is offered in spring only, and may be substituted in place of both CHEM 233 and CHEM 234.

**Chemistry BA - Additional Required Courses**

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<tr>
<th>Units</th>
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<tr>
<td>PHYS 115 or PHYS 121</td>
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<td>PHYS 116 or PHYS 122</td>
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<td>MATH 125 or MATH 121</td>
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<tr>
<td>MATH 126 or MATH 122</td>
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<td>Total Units</td>
<td>16</td>
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</table>

**Bachelor of Arts in Chemical Biology**

The BA program in chemical biology is intended for pre-professional students who plan careers in medicine, dentistry, veterinary medicine, pharmacy, or for individuals seeking careers that utilize chemistry to solve problems affecting living systems. A key component of the major is the flexibility imparted by fewer required courses and the integration of six credit hours of technical electives. Many chemical biology BA majors participate in undergraduate research within the Department of Chemistry (CHEM 397 / CHEM 398) or in other science departments, including those in the medical school.

**Total Units Required for Graduation:** 120

**Chemical Biology BA - Required Chemistry Courses**

<table>
<thead>
<tr>
<th>Units</th>
<th>Fall</th>
<th>Spring</th>
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<tbody>
<tr>
<td>Principles of Chemistry I (CHEM 105)</td>
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<tr>
<td>Principles of Chemistry II (CHEM 106)</td>
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<td>Principles of Chemistry Laboratory (CHEM 113)</td>
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**Second Year**

<table>
<thead>
<tr>
<th>Units</th>
<th>Fall</th>
<th>Spring</th>
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<tbody>
<tr>
<td>Organic Chemistry I (CHEM 323) or CHEM 223 and CHEM 224</td>
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<tr>
<td>Introductory Organic Chemistry Laboratory I (CHEM 233) (see below*) or Laboratory Methods in Organic Chemistry (CHEM 322)</td>
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<tr>
<td>Biochemistry Laboratory (CHEM 306)</td>
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<td></td>
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<tr>
<td>Introductory Biochemistry I (CHEM 328)</td>
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**Third Year**

<table>
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<th>Units</th>
<th>Fall</th>
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<tbody>
<tr>
<td>Introductory Physical Chemistry I (CHEM 301) or Physical Chemistry I (CHEM 335)</td>
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</tr>
<tr>
<td>Quantitative Analysis Laboratory (CHEM 304)</td>
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<tr>
<td>Foundations of Analytical Chemistry (CHEM 310)</td>
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<tr>
<td>Year Total:</td>
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**Fourth Year**

<table>
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<tr>
<th>Units</th>
<th>Fall</th>
<th>Spring</th>
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<tbody>
<tr>
<td>Technical Electives (see text below)</td>
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<tr>
<td>Undergraduate Research/Senior Capstone Project (CHEM 398)</td>
<td>3 - 6</td>
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<td>3-6</td>
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</tbody>
</table>

**Total Units in Sequence:** 36-43

* CHEM 322 is offered in spring only, and may be substituted in place of both CHEM 233 and CHEM 234. Only one semester of organic chemistry laboratory is required for our chemical biology BA program. However, some medical schools require two semesters of organic lab, so students should plan accordingly.

The technical electives may be chosen more widely from any of the physical sciences, math, or engineering courses. A maximum of six units of CHEM 397 may be taken as technical electives. Further additional units of CHEM 397 may be taken as free electives. Students may wish to group their electives into "tracks" of specialization in order to tailor their degree to a particular area of chemistry.


**Chemical Biology BA - Additional Required Courses**

<table>
<thead>
<tr>
<th>Course</th>
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<tbody>
<tr>
<td>BIOL 214</td>
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<tr>
<td>BIOL 215</td>
<td>Cells and Proteins</td>
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<td>Cells and Proteins Laboratory</td>
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<tr>
<td>PHYS 115</td>
<td>Introductory Physics I</td>
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<td>PHYS 116</td>
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<td>MATH 125</td>
<td>Math and Calculus Applications for Life</td>
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<td>or MATH 121</td>
<td>Calculus for Science and Engineering I</td>
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<td>MATH 126</td>
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<tr>
<td>or MATH 122</td>
<td>Calculus for Science and Engineering I</td>
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**Total Units** 24

**Departmental Honors**

Chemistry majors who have excellent academic records may participate in the Honors in Chemistry program. To graduate with honors in chemistry, a student must satisfy the following requirements:

1. A combined grade point average of 3.50 in chemistry, physics, and mathematics and an overall grade point average of 3.20
2. A minimum of six units of Undergraduate Research (CHEM 397), or chemical research done under another course number with departmental approval
3. A thesis approved by the department’s undergraduate affairs committee based on the level of research, quality of the manuscript, and chemical content

**Teacher Licensure in Chemistry**

The chemistry department offers a special option for undergraduate students who wish to pursue a chemistry major and a career in teaching. The Adolescent to Young Adult (AYA) Teacher Education Program in Physical Sciences prepares CWRU students to receive an Ohio Teaching License for grades 7-12. Students declare a second major in education—which involves 36 hours in education and practicum requirements—and complete a planned sequence of chemistry content course work within the context of the BA chemistry major. The program is designed to offer several unique features not found in other programs and to place students in mentored teaching situations throughout their teacher preparation career. This small, rigorous program is designed to capitalize on the strengths of CWRU’s chemistry department, its Teacher Education Program, and the relationships the university has built with area schools.

**Chemistry Minor**

Students may complete a minor in chemistry, defined as one year of freshman chemistry (including laboratory); two additional three-unit lecture courses; and two additional laboratory or approved courses. A recommended sequence would include:

**Course List**
Efforts are being made to understand the basic chemical properties leading to reactive mediators generated from physiological lipids.

Other research is aimed at developing new drugs for photodynamic therapy and at understanding the mechanism of action of drugs for antiretroviral therapy. The influence of metal ions in modifying reactivity is a common interest of several members of the faculty, as is the development of organometallic compounds for materials and catalysis. Chemical surfaces are being studied, and as are various applications of nanoparticles, from cells to the environment. Studies designed to characterize electrode-electrolyte interfaces, the electrochemical properties of new semiconductors, and single-cell microelectrodes are also ongoing. These efforts are complemented by theoretical studies on the interfacial structure and bonding of composite materials.

Case Western Reserve University ranks among the leading universities internationally in its strengths in electrochemistry and has brought these strengths together in the Yeager Center for Electrochemical Studies (YCES) (http://chemistry.case.edu/department/research/yces). The interdisciplinary nature of electrochemistry involves the interaction of electrochemists in the chemistry and chemical engineering departments with metallurgists, surface physicists, inorganic and organic chemists, polymer membrane chemists, and electrical engineers. Such interactions, lacking on most campuses, are promoted at Case Western Reserve University through YCES. Graduate students in the chemistry department have the opportunity to specialize in electrochemistry in one of the most extensive course and research programs in the United States.

Colloquia and Seminars

The department sponsors a rich program of colloquia and seminars on recent advances in chemical research. Most notable among these is the Frontiers in Chemistry Lecture Series, in which scientists of international distinction lecture on major discoveries and developments in chemistry. In addition, a weekly colloquium series provides lectures by invited speakers in a variety of fields of chemical investigation. Both of these programs are addressed to an audience of faculty, graduate students, and other chemical scientists in the university and the Cleveland area, and are a vital means to broaden current knowledge. Numerous other seminars and meetings are held on a more specialized and informal level. Most individual research groups conduct weekly discussions to evaluate their progress.

Courses

CHEM 105. Principles of Chemistry I. 3 Units.
Atomic structure; thermochemistry; periodicity, bonding and molecular structure; intermolecular forces; properties of solids; liquids, gases and solutions. Recommended preparation: One year of high school chemistry.

CHEM 106. Principles of Chemistry II. 3 Units.
Thermodynamics, chemical equilibrium; acid/base chemistry; oxidation and reduction; kinetics; spectroscopy; introduction to nuclear, organic, inorganic, and polymer chemistry. Prereq: CHEM 105 or CHEM 111.

CHEM 111. Principles of Chemistry for Engineers. 4 Units.
A first course in university chemistry emphasizing chemistry of materials for engineering students. Atomic theory and quantitative relationships; gas laws and kinetic theory; solutions, acid-base properties and pH; thermodynamics and equilibrium; kinetics, catalysis, and mechanisms; molecular structure and bonding. Recommended preparation: One year of high school chemistry.

CHEM 113. Principles of Chemistry Laboratory. 2 Units.
A one semester laboratory based on quantitative chemical measurements. Experiments include analysis, synthesis and characterization, thermochemistry and chemical kinetics. Computer analysis of data is a key part of all experiments. Prereq or Coreq: CHEM 105 or CHEM 106 or CHEM 111 or ENGR 145.

CHEM 119. Concepts for a Molecular View of Biology I. 3 Units.
The first semester of a two-course sequence in elementary inorganic, organic, and biochemistry, intended for nursing students or non-majors. Topics include: atomic theory, the periodic table, chemical bonds, molecular geometry, ideal gas laws, equilibrium and reaction rates, acids and bases, nuclear chemistry, and nomenclature and reactions of organic compounds (including alkyl, aryl, alcohol, carbonyl, and amino compounds). Problems involving numeric computation are emphasized. This course is not open to students with credit for CHEM 105 or CHEM 111.

CHEM 121. Concepts for a Molecular View of Biology II. 3 Units.
The second course of a two-semester sequence in elementary inorganic, organic, and biochemistry, intended for nursing students or non-majors. Topics include: carbohydrates, lipids, proteins, enzyme kinetics, metabolic pathways and bioenergetics, DNA and RNA, methods of molecular biology, and nutrition. Applications to human physiology and medicine emphasized. This course is not open to students with credit for CHEM 223 or CHEM 323. Prereq: CHEM 119.

CHEM 223. Introductory Organic Chemistry I. 3 Units.
Introductory course for science majors and engineering students. Develops themes of structure and bonding along with elementary reaction mechanisms. Includes treatment of hydrocarbons, alkyl halides, alcohols, and ethers as well as an introduction to spectroscopy. Prereq: CHEM 106 or ENGR 145.

CHEM 224. Introductory Organic Chemistry II. 3 Units.
Continues and extends themes of structure and bonding from CHEM 223 and continues spectroscopy and more complex reaction mechanisms. Includes treatment of aromatic rings, carbonyl compounds, amines, and selected special topics. Prereq: CHEM 223 or CHEM 323.

CHEM 233. Introductory Organic Chemistry Laboratory I. 2 Units.
An introductory organic laboratory course emphasizing microscale operations. Synthesis and purification of organic compounds, isolation of natural products, and systematic identification of organic compounds by physical and chemical methods. Prereq: (CHEM 106 or ENGR 145) and CHEM 113. Prereq or Coreq: CHEM 223 or CHEM 323.

CHEM 234. Introductory Organic Chemistry Laboratory II. 2 Units.
A continuation of CHEM 233, involving multi-step organic synthesis, peptide synthesis, product purification and analysis using sophisticated analytical techniques such as chromatography and magnetic resonance spectroscopy. Prereq: CHEM 233. Prereq or Coreq: CHEM 224

CHEM 290. Chemical Laboratory Methods for Engineers. 3 Units.
Techniques of chemical synthesis, analysis, and characterization. Uses students’ backgrounds in general and organic chemistry, but requires no background in chemical laboratory operations. Prereq or Coreq: CHEM 223 or CHEM 323.
CHEM 301. Introductory Physical Chemistry I. 3 Units.
First of a two-semester sequence covering principles and applications of physical chemistry, intended for chemistry and engineering majors and other students having primary interests in biochemical, biological or life-sciences areas. States and properties of matter. Thermodynamics and its application to chemical and biochemical systems. Chemical equilibrium. Electrochemistry. Recommended preparation: One year each of undergraduate physics and calculus, preferably including partial derivatives. Prereq: CHEM 106.

CHEM 302. Introductory Physical Chemistry II. 3 Units.

CHEM 304. Quantitative Analysis Laboratory. 2 Units.
A one-semester laboratory course providing practical experience in the analytical process. Focus is on statistical error analysis of measurements, method validation and instrument calibration, and reporting. Basic laboratory skills are developed and evaluated based on accuracy and precision of measurements. Prereq: CHEM 106 and CHEM 113. Coreq: CHEM 310.

CHEM 305. Introductory Physical Chemistry Laboratory. 3 Units.
A one-semester laboratory course focusing on the principles and quantitative characterization of chemical and biochemical systems. Experiments include chemical equilibrium kinetics, electrochemistry, spectroscopy and the use of computers for the statistical analysis of experimental data. Seminar discussions and disciplinary writing of results. Counts as SAGES Departmental Seminar. Prereq: CHEM 301 and CHEM 304 or CHEM 335. Or Prereq or Coreq: CHEM 302 or CHEM 336.

CHEM 306. Biochemistry Laboratory. 3 Units.
A one-semester laboratory and lecture course developed to introduce students to a variety of chemical biology laboratory themes including buffering, identification of amino acids, immunosassay, ligand binding, cellular fractionation, enzyme isolation and purification, proteomics, and enzyme kinetics. Techniques include titration, various forms of chromatography, colorimetric assays, electrophoresis, high performance liquid chromatography and liquid chromatography coupled with tandem mass spectrometry. Recommended preparation: CHEM 328/428. Counts as SAGES Departmental Seminar. Prereq: CHEM 233.

CHEM 310. Foundations of Analytical Chemistry. 3 Units.
A one-semester lecture covering classical and modern aspects of the analytical process; analysis requirements, method selection including capabilities and limitations, sampling and sample processing, measurement data statistics for evaluation of precision and accuracy, method validation, and reporting. Fundamental concepts in equilibrium thermodynamics are covered in the context of chemical analysis. Methods based on titration, spectroscopy, electrochemistry, chromatography, and mass spectrometry are emphasized. Prereq: CHEM 106 and CHEM 113. Coreq: CHEM 304.

CHEM 311. Inorganic Chemistry I. 3 Units.
Fundamentals of inorganic chemistry. Topics include molecular structure, molecular shape and symmetry, structure of solids, d-metal complexes, oxidation and reduction, and acids and bases. Prereq or Coreq: CHEM 301 or CHEM 335.

CHEM 316. Frontiers of Inorganic Chemistry. 3 Units.
This course deals with five topics in inorganic chemistry of current interest. The topics are: ways in which inorganic chemistry can increase the quality of the environment, methods by which inorganic chemistry can lead to sustainable processes in a developed industrial society, advances in bioinorganic and medicinal inorganic chemistry of clinical importance, modern inorganic materials with unusual and valuable property sets, and representative industrial inorganic research and production processes. It is to be team taught. Offered as CHEM 316 and CHEM 416.

CHEM 322. Laboratory Methods in Organic Chemistry. 3 Units.
Experimental approach to the synthesis, purification and characterization of organic compounds. Nuclear magnetic resonance (NMR) and infrared (IR) spectroscopies; chromatographic techniques. Prereq: CHEM 304 and CHEM 223 or CHEM 323. Prereq or Coreq: CHEM 224 or CHEM 324.

CHEM 323. Organic Chemistry I. 3 Units.
Relationships between molecular structure and chemical reactivity and development of sophisticated problem-solving skills in the context of organic reaction mechanisms and multi-step synthesis. Homolysis and heterolytic substitution, elimination, oxidation and reduction reactions; topics in stereochemistry and spectroscopy. Recommended for chemistry, biochemistry, and related majors. Prereq: CHEM 106 or ENGR 145.

CHEM 324. Organic Chemistry II. 3 Units.
Continuation of CHEM 323. Introduces the chemistry of carbonyl, aromatic and amino functional groups, and develops the concepts of conjugation and resonance, molecular orbital theory and pericyclic reactions. Prereq: CHEM 223 or CHEM 323.

CHEM 325. Physical Methods for Determining Organic Structure. 3 Units.
Structure determination of organic compounds using mass spectrometry and modern instrumental techniques such as infrared, ultraviolet, visible, and nuclear magnetic resonance spectroscopy. Recommended preparation: Two semesters of undergraduate organic chemistry. Offered as CHEM 325 and CHEM 425.

CHEM 328. Introductory Biochemistry I. 3 Units.

CHEM 329. Biochemistry II: Living Systems. 3 Units.

CHEM 331. Laboratory Methods in Inorganic Chemistry. 3 Units.
Synthesis, separation techniques, physical properties, and analysis. Advanced techniques of chemical synthesis, leading the student to the preparation of interesting inorganic and organometallic compounds. Offered as: CHEM 331 and CHEM 431. Prereq: CHEM 322.
CHEM 332. Laboratory Methods in Physical Chemistry. 3 Units.

CHEM 333. Medicinal Chemistry and Drug Development. 3 Units.
This course provides an overview on how principles in chemistry and biology are integrated to facilitate drug development. Primary emphasis will be placed on the development of organic molecules as drugs and metabolic enzymes as drug targets. Subjects pertinent to the introduction of medicinal chemistry, evaluation of drug efficacies in vitro and in vivo, and drug metabolism will be covered. Offered as CHEM 333 and CHEM 433. Prereq: CHEM 223 or CHEM 323 and BIOL 215. Coreq: CHEM 224 or CHEM 324.

CHEM 335. Physical Chemistry I. 3 Units.
First of a two-semester sequence of physical chemistry for chemistry majors and others with career goals in the physical sciences or engineering. Thermodynamics and its application to chemical systems: First Law, Heat, Work, internal energy, State functions, Thermochromy, Entropy, 2nd and 3rd law of thermodynamics, Chemical equilibrium, Real Gases, Phase diagrams, ideal and real solutions, Electrolyte solutions, and Electrochemical cells, batteries and fuel cells. Introduction to chemical kinetics. This class is taught using the flipped classroom strategy, where students are expected to learn the basic concepts before class by reading relevant notes and textbook sections, as well as watching videos and answering pre-class questions. In class, students work on exercises and solving problems. Students have ample opportunities to clarify concepts, ask questions, and learn from both peers (in group settings) and the teacher. Recommended preparation: One year each of undergraduate physics and calculus, including partial derivatives. Prereq: CHEM 106 or ENGR 145.

CHEM 336. Physical Chemistry II. 3 Units.

CHEM 337. Quantum Mechanics I. 3 Units.
Introduction to quantization, measurement and the Schrodinger equation; angular momentum and states of molecules. Perturbation theory, spectroscopy and chemical bonding. Variational theory and calculations of molecular properties. Offered as CHEM 335 and CHEM 446. Prereq: CHEM 336.

CHEM 340. Solar Energy Conversion. 3 Units.
This is a multidisciplinary course from a chemist’s point of view. This course teaches the background necessary to read and understand the scientific literature on solar energy conversion, and includes some basic device physics, materials chemistry and chemistry. Topics provide an overview of the field and includes: Global energy perspective, principles of photovoltaics, crystalline solar cells, thin-film solar cells, dye-sensitized solar cells, organic solar cells (with emphasis on polymer-based solar cells), photoelectrochemical cells and artificial photosynthesis for fuel production, and semiconductor nanostructures and quantum dots for solar energy conversion. The course includes three laboratories and a demo using state-of-the-art equipment, as well as presentations of recent research articles by the graduate students. It is recommended that students have experience with thermodynamics. The following CWRU courses would meet this expectation: CHEM 301, CHEM 335, ENGR 225 or PHYS 313. Offered as CHEM 340 and CHEM 440. Prereq: CHEM 106 or ENGR 145.

CHEM 341. Functional Nanomaterials. 3 Units.
This course is designed to introduce important concepts on the fundamental physical and chemical properties of technologically important nanometer scale materials. The course will cover an overview of the scientific principles pertaining to new properties at the nanoscale; synthesis and characterization tools; and existing and emerging applications of nanomaterials. It will center on current research developments on major classes of functional nanomaterials, including plasmonic nanoparticles, quantum dots, nanomagnets, carbon nanotubes, nanocatalysts and hybrid inorganic/organic nanostructures. In addition an emphasis will be placed on understanding the broader societal, economical and environmental impact of the scientific and technological advances brought forward by nanotechnology. Offered as CHEM 341 and CHEM 441.

CHEM 342. Computational Chemistry. 3 Units.
An introduction to computational methods in electronic structure. Molecular mechanics, semiempirical molecular orbital calculations, ab initio, post Hartree-Fock, density-functional theories, and hybrid approaches will be addressed. Continuum solvation calculations will be considered, time permitting. Offered as CHEM 342 and CHEM 442. Prereq: CHEM 223 or CHEM 323.

CHEM 344. The Chemistry and Physics of Energy Storage. 3 Units.
This course will cover both scientific and economic aspects of the operation of energy storage devices currently being considered for both small and large scale applications ranging from portable electronics to the electrical grid. These devices include pumped hydro, flywheel, compressed air, batteries, supercapacitors, thermal conversion, regenerative fuel cells and redox flow cells. Not to be included in this course are energy conversion devices such as photovoltaics and windmills. This course would be of interest to both undergraduate and graduate students with interest in the general area of energy management and will cover the physics and chemistry principles associated with the various modes of storage. Students either individually or in small groups will be expected to prepare a written document at the end of the course that describes and summarizes each mode of storage, including a discussion of all aspects of the technology such as costs of installation and operation, environmental impact, and economic projections. As part of this exercise students will become familiar with the extraordinary resources offered by our library. Offered as CHEM 344 and CHEM 444. Prereq: CHEM 106.

CHEM 348. Chemistry Fermentation Brewing: To Discern the Molecular Basis of Fermentation and Beer Brewing. 3 Units.
This course includes in-depth discussions of the chemistry and procedures underlying either an aspect of the brewing process or a style of beer in order to discern the molecular basis of fermentation and beer brewing. The biochemistry of yeast fermentation and hops, as well as mashing, lautering, boiling, conditioning, filtering, and packaging will be discussed. There is no lab component (such as brewing beer), although field trips to pubs including the Jolly-Scholar pub (located on campus) will be part of the course, as well as invited speakers who have set up local microbreweries. Each student will be expected to have basic background knowledge of chemistry, such as material taught in standard first year General Chemistry courses. Offered as CHEM 348 and CHEM 448. Prereq: CHEM 106 or ENGR 145.

CHEM 395. Chemistry Colloquium Series. 1 Unit.
Course content provided by Thursday chemistry department colloquia (or Frontiers in Chemistry lectures). Discussion sessions review previous lectures and lay foundation for forthcoming lectures.
CHEM 397. Undergraduate Research. 1 - 6 Units.
Independent research project within a research group in the chemistry department or, by petition, within a research group in another Case department. Arrangements should be made with the faculty member selected. Open to all chemistry majors and other qualified students; required for Honors in Chemistry. A written report is required each semester.

CHEM 398. Undergraduate Research/Senior Capstone Project. 3 - 6 Units.
Independent research project within a research group in the chemistry department or, by petition, within a research group in another Case department. Arrangements should be made by consultation with the faculty member selected and the Senior Capstone Committee of the chemistry department. Open to all chemistry majors and other qualified students. Satisfies the research requirement for Honors in Chemistry. A written report and public oral presentations are required. Counts as SAGES Senior Capstone.

CHEM 406. Chemical Kinetics. 3 Units.
Theory and characterization of chemical rate processes. Recommended preparation: Two semesters of undergraduate physical chemistry.

CHEM 407. Chemical Thermodynamics. 3 Units.
Thermodynamics and statistical thermodynamics and their application to chemical problems. Recommended preparation: Two semesters of undergraduate physical chemistry.

CHEM 412. Advanced Inorganic Chemistry I. 3 Units.
Chemistry of inorganic systems. Spectroscopy, magnetism, and stereochemistry of transition metal compounds. Recommended preparation: One semester of undergraduate inorganic chemistry and two semesters of undergraduate physical chemistry.

CHEM 414. Organometallic Reactions and Structures. 3 Units.
Bonding, structure, and mechanistic aspects of organometallic chemistry and the relevance of organometallic species to chemical catalysis. Recommended preparation: One semester of undergraduate inorganic chemistry.

CHEM 416. Frontiers of Inorganic Chemistry. 3 Units.
This course deals with five topics in inorganic chemistry of current interest. The topics are: ways in which inorganic chemistry can increase the quality of the environment, methods by which inorganic chemistry can lead to sustainable processes in a developed industrial society, advances in bioinorganic and medicinal inorganic chemistry of clinical importance, modern inorganic materials with unusual and valuable property sets, and representative industrial inorganic research and production processes. It is to be team taught. Offered as CHEM 316 and CHEM 416.

CHEM 421. Advanced Organic Chemistry I. 3 Units.

CHEM 422. Advanced Organic Chemistry II. 3 Units.

CHEM 425. Physical Methods for Determining Organic Structure. 3 Units.
Structure determination of organic compounds using mass spectrometry and modern instrumental techniques such as infrared, ultraviolet, visible, and nuclear magnetic resonance spectroscopy. Recommended preparation: Two semesters of undergraduate organic chemistry. Offered as CHEM 325 and CHEM 425.

CHEM 428. Introductory Biochemistry I. 3 Units.

CHEM 429. Biochemistry II: Living Systems. 3 Units.

CHEM 431. Laboratory Methods in Inorganic Chemistry. 3 Units.
Synthesis, separation techniques, physical properties, and analysis. Advanced techniques of chemical synthesis, leading the student to the preparation of interesting inorganic and organometallic compounds. Offered as CHEM 331 and CHEM 431. Prereq: CHEM 322

CHEM 433. Medicinal Chemistry and Drug Development. 3 Units.
This course provides an overview on how principles in chemistry and biology are integrated to facilitate drug development. Primary emphasis will be placed on the development of organic molecules as drugs and metabolic enzymes as drug targets. Subjects pertinent to the introduction of medicinal chemistry, evaluation of drug efficacies in vitro and in vivo, and drug metabolism will be covered. Offered as CHEM 333 and CHEM 433.

CHEM 435. Synthetic Methods in Organic Chemistry. 3 Units.

CHEM 436. Complex Molecular Synthesis. 3 Units.
An advanced organic chemistry course providing students with an in-depth examination of the art of total synthesis drawing from both classical and recent examples. Recommended preparation: Two semesters of undergraduate organic chemistry.
CHEM 440. Solar Energy Conversion. 3 Units.
This is a multidisciplinary course from a chemist's point of view. This course teaches the background necessary to read and understand the scientific literature on solar energy conversion, and includes some basic device physics, materials chemistry and chemistry. Topics provide an overview of the field and include: Global energy perspective, principles of photovoltaics, crystalline solar cells, thin-film solar cells, dye-sensitized solar cells, organic solar cells (with emphasis on polymer-based solar cells), photoelectrochemical cells and artificial photosynthesis for fuel production, and semiconductor nanostructures and quantum dots for solar energy conversion. The course includes three laboratories and a demo using state-of-the-art equipment, as well as presentations of recent research articles by the graduate students. It is recommended that students have experience with thermodynamics. The following CWRU courses would meet this expectation: CHEM 301, CHEM 335, ENGR 225 or PHYS 313. Offered as CHEM 340 and CHEM 440.

CHEM 441. Functional Nanomaterials. 3 Units.
This course is designed to introduce important concepts on the fundamental physical and chemical properties of technologically important nanometer scale materials. The course will cover an overview of the scientific principles pertaining to new properties at the nanoscale; synthesis and characterization tools; and existing and emerging applications of nanomaterials. It will center on current research developments on major classes of functional nanomaterials, including plasmonic nanoparticles, quantum dots, nanomagnets, carbon nanotubes, nanocatalysts and hybrid inorganic/organic nanostructures. In addition an emphasis will be placed on understanding the broader societal, economical and environmental impact of the scientific and technological advances brought forward by nanotechnology. Offered as CHEM 341 and CHEM 441.

CHEM 442. Computational Chemistry. 3 Units.
An introduction to computational methods in electronic structure. Molecular mechanics, semiempirical molecular orbital calculations, ab initio, post Hartree-Fock, density-functional theories, and hybrid approaches will be addressed. Continuum solvation calculations will be considered, time permitting. Offered as CHEM 342 and CHEM 442. Prereq: CHEM 223 or CHEM 323.

CHEM 444. The Chemistry and Physics of Energy Storage. 3 Units.
This course will cover both scientific and economic aspects of the operation of energy storage devices currently being considered for both small and large scale applications ranging from portable electronics to the electrical grid. These devices include pumped hydro, flywheel, compressed air, batteries, supercapacitors, thermal conversion, regenerative fuel cells and redox flow cells. Not to be included in this course are energy conversion devices such as photovoltaics and windmills. This course would be of interest to both undergraduate and graduate students with interest in the general area of energy management and will cover the physics and chemistry principles associated with the various modes of storage. Students either individually or in small groups will be expected to prepare a written document at the end of the course that describes and summarizes each mode of storage, including a discussion of all aspects of the technology such as costs of installation and operation, environmental impact, and economic projections. As part of this exercise students will become familiar with the extraordinary resources offered by our library. Offered as CHEM 344 and CHEM 444. Prereq: CHEM 106.

CHEM 445. Electrochemistry I. 3 Units.
Electrochemical properties and processes of electrode/electrolyte interfaces. Fundamental background for work in corrosion, electrodeposition, industrial electrolysis, electro-organic synthesis, batteries, fuel cells, and photoelectrochemical energy conversion. Recommended preparation: One semester of undergraduate physical chemistry.

CHEM 446. Quantum Mechanics I. 3 Units.
Introduction to quantization, measurement and the Schroedinger equation; angular momentum and states of molecules. Perturbation theory, spectroscopy and chemical bonding. Variational theory and calculations of molecular properties. Recommended preparation: Two semesters of undergraduate physical chemistry. Offered as CHEM 335 and CHEM 446.

CHEM 448. Chemistry Fermentation Brewing: To Discern the Molecular Basis of Fermentation and Beer Brewing. 3 Units.
This course includes in-depth discussions of the chemistry and procedures underlying either an aspect of the brewing process or a style of beer in order to discern the molecular basis of fermentation and beer brewing. The biochemistry of yeast fermentation and hops, as well as mashing, lautering, boiling, conditioning, filtering, and packaging will be discussed. There is no lab component (such as brewing beer), although field trips to pubs including the Jolly-Scholar pub (located on campus) will be part of the course, as well as invited speakers who have set up local microbreweries. Each student will be expected to have basic background knowledge of chemistry, such as material taught in standard first year General Chemistry courses. Offered as CHEM 348 and CHEM 448. Prereq: CHEM 106 or ENGR 145.

CHEM 450. Molecular Spectroscopy. 3 Units.
Translation, rotation, vibration, and electronic transitions of molecules. Prereq: CHEM 446.

CHEM 475. Protein Biophysics. 3 Units.
This course focuses on in-depth understanding of the molecular biophysics of proteins. Structural, thermodynamic and kinetic aspects of protein function and structure-function relationships will be considered at the advanced conceptual level. The application of these theoretical frameworks will be illustrated with examples from the literature and integration of biophysical knowledge with description at the cellular and systems level. The format consists of lectures, problem sets, and student presentations. A special emphasis will be placed on discussion of original publications. Offered as BIQC 475, CHEM 475, PHOL 475, PHRM 475, and NEUR 475.

CHEM 491. Modern Chemistry for Innovation I. 3 Units.
The first half of a two-semester sequence providing an understanding of chemistry as a basis for successfully launching new high-tech ventures. The course will examine physical limitations to present technologies and the use of chemistry to identify potential opportunities for new venture creation. The course will provide experience in using chemistry for both identification of incremental improvements and as the basis for alternative technologies. Case studies will be used to illustrate recent commercially successful (and unsuccessful) venture creation and will illustrate characteristics for success.
CHEM 493. Feasibility and Technology Analysis. 3 Units.
This course provides the tools scientists need to determine whether a technology is ready for commercialization. These tools include (but are not limited to): financial analysis, market analysis, industry analysis, technology analysis, intellectual property protection, the entrepreneurial process and culture, an introduction to entrepreneurial strategy and new venture financing. Deliverables will include a technology feasibility analysis on a possible application in the student’s scientific area. Offered as BIOL 493, CHEM 493, and PHYS 493.

CHEM 502. Special Topics in Inorganic Chemistry. 1 - 6 Units.
(Credit as arranged.) Lectures on advanced topics in inorganic chemistry presented by staff or visiting lecturers. Course title, content, and credit change from year to year.

CHEM 506. Special Topics in Physical Chemistry. 1 - 6 Units.
(Credit as arranged.) Lectures on advanced topics in physical chemistry presented by staff or visiting lecturers. Course title, content, and credit change from year to year.

CHEM 507. Special Readings in Chemistry. 1 - 6 Units.
Detailed study of a special topic in chemistry under the guidance of a faculty member.

CHEM 508. Special Readings in Chemistry. 1 - 6 Units.
Detailed study of a special topic in chemistry under the guidance of a faculty member.

CHEM 601. Research. 1 - 18 Units.
(Credit as arranged.) Special research in an area of chemistry under the guidance of a faculty member.

CHEM 605. Chemistry Colloquium Series. 0 Unit.
Course content provided by Thursday chemistry department colloquia (or Frontiers in Chemistry lectures). Discussion sessions review previous lectures and lay foundation for forthcoming lectures.

CHEM 651. Thesis M.S.. 1 - 18 Units.
(Credit as arranged.)

CHEM 701. Dissertation Ph.D.. 1 - 9 Units.
(Credit as arranged.) Prereq: Predoctoral research consent or advanced to Ph.D. candidacy milestone.